

Localization-Delocalization Matrices (LDMs) and Electron Density-Weighted Adjacency Matrices (EDWAMs): A Bridge between the Quantum Theory of Atoms in Molecules and Chemical Graph Theory

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The quantum theory of atoms in molecules (QTAIM) [1] defines localization and delocalization indices (LIs ($\Lambda(\Omega)$), and DIs ($\delta(\Omega, \Omega')$)) in terms of the double integral of the Fermi hole density over a single atomic basin or over two basins, respectively [2]. These indices, arranged in a “localization-delocalization matrix (LDM, or ζ -matrix)”, can be treated by the tools of chemical graph theory [3-6] to extract graph invariants, *e.g.* eigenvalues or matrix distances from a given reference, for use into QSAR [7-15]. The LDM is defined [7]:

$$\zeta \equiv \begin{matrix} \begin{matrix} \Lambda(\Omega_1)\delta(\Omega_1, \Omega_1) & \delta(\Omega_1, \Omega_2) & \delta(\Omega_1, \Omega_n) \\ \delta(\Omega_2, \Omega_1)/2 & \Lambda(\Omega_2) & \delta(\Omega_2, \Omega_n) \\ \text{M} & \text{M} & \text{O} \\ \delta(\Omega_n, \Omega_1)/2 & \delta(\Omega_n, \Omega_2)/2 & \Lambda(\Omega_n) \end{matrix} & \begin{matrix} \delta(\Omega_1, \Omega_1) \\ \delta(\Omega_2, \Omega_2) \\ \text{M} \\ \delta(\Omega_n, \Omega_n) \end{matrix} & \begin{matrix} = N(\Omega_1) \\ = N(\Omega_2) \\ \text{M} \\ = N(\Omega_n) \end{matrix} \\ \sum_{\text{column}} & \sum_{i=1}^n N(\Omega_i) = N & \end{matrix}$$

$$\sum_{i=1}^n N(\Omega_i) = N \quad \text{tr}(\zeta) = N_{\text{loc}}$$

The example in Fig. 1 shows the correlation between the matrix distance of the diagonal-suppressed LDM (DM) of partial molecular graphs with pK_a of a series of benzoic acids.

A matrix representative of the QTAIM molecular graph can also be constructed by inserting the electron density at the bond critical point (ρ_{BCP}) as the matrix element common between two bonded atoms. This matrix, termed “electron density-weighted adjacency matrix (EDWAM)” [7], is accessible from X-ray diffraction experiments, unlike the LDM. These matrix representatives of molecules provide the basis for robust empirical correlations and predictions of properties including pK_a 's, λ_{max} 's, aromaticity of rings in polycyclic aromatic hydrocarbons, corrosion inhibition efficiencies, $\log P$'s, $\log \text{IC}_{50\%}$, and provide a tool for assessing the quality of basis sets in electronic structure calculations. The same tools developed for LDMs are straightforwardly extendable to the “Interacting Quantum Atoms (IQA)” energy components [16].

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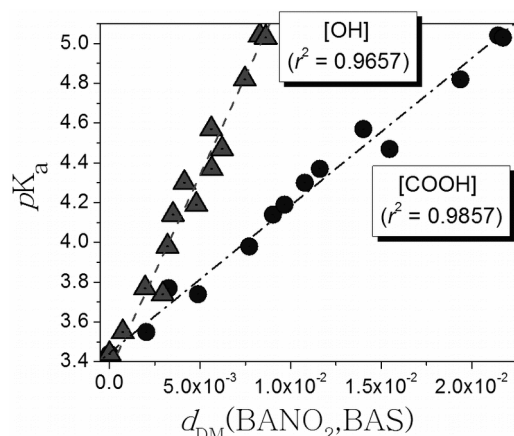


Fig. 1. Correlations between experimental pK_a and Frobenius distance of the partial delocalization matrices (DMs) from the most acidic molecule in the group (*p*-nitrobenzoic acid, or BA-NO₂). The partial DMs for the least steep plot include the atoms of the -COOH group while those of the steepest plot include the -OH group only. (From Ref. [12]).